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3-Methoxycarbonyl-1-methyl-4-trifluoromethylpyrazole †

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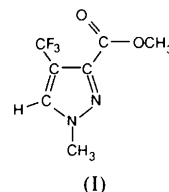
Abstract

The crystallographic characterization of the title molecule, $C_7H_7F_3N_2O_2$, distinguishes it from its isomer, 5-methoxycarbonyl-1-methyl-4-trifluoromethylpyrazole, which is the major product in their joint synthesis. The bond lengths, except for C3—C4, are consistent with other molecules having substituents on the same ring positions. The pyrazole ring and the carboxy group plane are nearly coplanar.

† IUPAC name: methyl 1-methyl-4-trifluoromethyl-3-pyrazolecarboxylate.

Comment

The title molecule (I) and its 5-methoxycarbonyl isomer are obtained, along with other cyclic products, by the reaction of 4,4,4-trifluorobut-1-ynoic acid and diazomethane (Tajammal, 1988, 1991; Tajammal & Tipping, 1990). The title molecule was recrystallized from dichloromethane after chromatographic separation.



The bond lengths in the pyrazole ring are typical of pyrazole molecules with substituents at the same ring positions (Becher, Brondum, Krake, Pluta, Simonsen, Molina & Begtrup, 1988; Cousson, Robert & Hubert-Habart, 1991; Lapasset & Falgueirettes, 1972), except for C3—C4 which is up to 0.05 Å shorter.

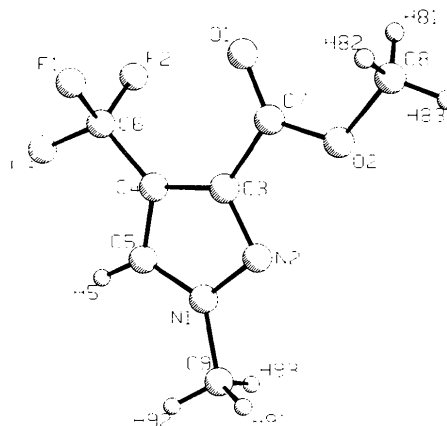


Fig. 1. View of the title molecule showing the atom-labelling scheme.

Experimental

Crystal data

$C_7H_7F_3N_2O_2$
 $M_r = 208.14$
Monoclinic
 $P2_1/c$
 $a = 4.624 (2) \text{ \AA}$
 $b = 16.600 (2) \text{ \AA}$
 $c = 11.687 (2) \text{ \AA}$
 $\beta = 92.43 (2)^\circ$
 $V = 896.3 (7) \text{ \AA}^3$
 $Z = 4$
 $D_x = 1.542 \text{ Mg m}^{-3}$

Data collection

Enraf-Nonius CAD-4
diffractometer

Mo $K\alpha$ radiation
 $\lambda = 0.71069 \text{ \AA}$
Cell parameters from 20
reflections
 $\theta = 6.35\text{--}11.3^\circ$
 $\mu = 0.146 \text{ mm}^{-1}$
 $T = 296 \text{ K}$
Block
 $0.3 \times 0.3 \times 0.2 \text{ mm}$
Colourless

$R_{\text{int}} = 0.045$
 $\theta_{\text{max}} = 25^\circ$

$\omega/2\theta$ scans $h = -4 \rightarrow 4$
 Absorption correction: $k = 0 \rightarrow 16$
 none $l = -11 \rightarrow 11$
 1284 measured reflections 3 standard reflections
 721 independent reflections frequency: 120 min
 522 observed reflections intensity variation: none
 $[I > 2\sigma(I)]$

Refinement

Refinement on F $(\Delta/\sigma)_{\max} = 0.04$
 $R = 0.071$ $\Delta\rho_{\max} = 0.33 \text{ e } \text{\AA}^{-3}$
 $wR = 0.040$ $\Delta\rho_{\min} = -0.30 \text{ e } \text{\AA}^{-3}$
 $S = 3.00$ Atomic scattering factors
 522 reflections from *International Tables*
 127 parameters for *X-ray Crystallography*
 H-atom parameters not refined (1974, Vol. IV)
 Weights as default in
 TEXSAN

Data collection: CAD-4 Software (Enraf-Nonius, 1989). Cell refinement: CAD-4 Software. Data reduction: SHELX76 (Sheldrick, 1976). Program(s) used to solve structure: MITHRIL (Gilmore, 1984). Program(s) used to refine structure: TEXSAN (Molecular Structure Corporation, 1985). Molecular graphics: PLUTO (Motherwell & Clegg, 1978). Software used to prepare material for publication: CIF (Hall, Allen & Brown, 1991).

Table 1. Fractional atomic coordinates and equivalent isotropic thermal parameters (\AA^2)

$$B_{\text{eq}} = (8\pi^2/3) \sum_i \sum_j U_{ij} a_i^* a_j^* \mathbf{a}_i \cdot \mathbf{a}_j$$

	x	y	z	B_{eq}
F1	0.4080 (13)	0.1649 (4)	0.6144 (5)	6.1 (4)
F2	0.0379 (13)	0.1554 (4)	0.7181 (6)	6.5 (4)
F3	-0.0007 (13)	0.1211 (4)	0.5448 (5)	6.9 (4)
O1	0.425 (2)	0.2980 (4)	0.7814 (7)	6.4 (5)
O2	0.329 (2)	0.4275 (5)	0.7495 (6)	5.8 (5)
N1	-0.189 (2)	0.3563 (6)	0.4962 (7)	4.0 (5)
N2	-0.028 (2)	0.3941 (5)	0.5784 (8)	4.2 (5)
C3	0.101 (2)	0.3325 (7)	0.6307 (9)	3.8 (7)
C4	0.029 (2)	0.2585 (7)	0.5834 (9)	3.7 (7)
C5	-0.159 (2)	0.2769 (6)	0.4951 (9)	4.0 (7)
C6	0.126 (3)	0.1765 (8)	0.6155 (11)	4.8 (8)
C7	0.303 (3)	0.3503 (8)	0.7312 (10)	4.6 (8)
C8	0.536 (3)	0.4523 (7)	0.8382 (10)	7.5 (9)
C9	-0.379 (2)	0.4066 (6)	0.4156 (9)	5.3 (7)

Table 2. Geometric parameters (\AA , $^\circ$)

F1—C6	1.32 (1)	N1—C5	1.33 (1)
F2—C6	1.33 (1)	N1—C9	1.51 (1)
F3—C6	1.35 (1)	N2—C3	1.32 (1)
O1—C7	1.18 (1)	C3—C4	1.38 (1)
O2—C7	1.30 (1)	C3—C7	1.50 (1)
O2—C8	1.44 (1)	C4—C5	1.35 (1)
N1—N2	1.35 (1)	C4—C6	1.48 (1)
C7—O2—C8	116.9 (9)	N1—C5—C4	106 (1)
N2—N1—C5	114.4 (9)	F1—C6—F2	108 (1)
N2—N1—C9	118.5 (9)	F1—C6—F3	107 (1)
C5—N1—C9	127 (1)	F1—C6—C4	115 (1)
N1—N2—C3	101.2 (8)	F2—C6—F3	103 (1)
N2—C3—C4	114.1 (9)	F2—C6—C4	112 (1)
N2—C3—C7	118 (1)	F3—C6—C4	111 (1)
C4—C3—C7	128 (1)	O1—C7—O2	127 (1)
C3—C4—C5	103.8 (9)	O1—C7—C3	121 (1)
C3—C4—C6	131 (1)	O2—C7—C3	112 (1)
C5—C4—C6	125 (1)		

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Lists of structure factors, anisotropic thermal parameters, H-atom coordinates and complete geometry including torsion angles have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71507 (8 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: HA1068]

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meso- and rac-2,7-Diphenyl-3,5-octadiyne-2,7-diol, C₂₀H₁₈O₂

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Abstract

The stereoisomeric forms of 2,7-diphenyl-3,5-octadiyne-2,7-diol (I) have been identified; meso- and rac-(I) exhibit different crystal packing, reflecting very different hydrogen-bonding networks.

Comment

Copper(I)-induced oxidative coupling of terminal acetylenes, the so-called Glaser reaction, provides a